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Teaching Transparency Chemistry Chapter 19

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TEACHING TRANSPARENCY WORKSHEET Ionic Bonds Use with Chapter 7, Section 7.2 1. How many valence electrons does a neutral magnesium (Mg) atom have? 2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl atom have? 4.

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a. potassium-39 b. potassium-40 c. potassium-41 7. Write an equation showing the relationship between an atom's atomic number and its mass number. 8. Lithium has two isotopes: lithium-6 and lithium-7. Draw a diagram, like those shown on the transparency, for each lithium isotope. Label the protons, electrons, neutrons, and

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