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is needed if the gas sample is going to have an internal pressure that is different from the external pressure. If a flexible container is used, the internal pressure. Page 3/11.
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**Gas Variables
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Model 1 - Gases in a
Nonflexible Container
Experiment A (Adding
more gas) A1 A2 A3

Volume = 1 unit

Volume = 1 unit

Volume = 1 unit

External pressure = 1

atm External pressure

= 1 atm External

pressure = 1 atm

Internal pressure = 1

atm Internal pressure

= 2 atm Internal

pressure = 3 atm

Temperature = 200 K

Temperature = 200 K

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Temperature = 200 K
Experiment B (Heating
the gas) B1 B2 B3
Volume = 1 unit
Volume = 1 unit
Volume = 1 unit
External pressure = 1
atm External pressure
= 1 atm ...

Gas Variables Nonflexible Container Answer Key

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Model 1 Gases in a
Nonflexible Container
Experiment A ... (V),

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temperature (T), and moles (n) of gas. These four variables can be related mathematically so that predictions about gas behavior can be made. Model 1 - Gases in a Nonflexible Container Experiment A (Adding more gas)
A1 A3A2

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A gas exerts a pressure

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of 3 kPa on the walls of container 1. When container 1 is emptied into a 10-liter container, the pressure exerted by the gas increases to 6 kPa. Find the volume of container 1. Assume that the temperature and quantity of the gas remain constant.
Given, Initial pressure, $P_1 = 3\text{kPa}$. Final pressure, $P_2 = 6\text{kPa}$. Final ...

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Nonflexible **Boyle's Law -**

Statement, Detailed Explanation, and Examples

Question: What will happen to the pressure inside a container of gas if some of the gas is allowed to escape?

Select one: a. there will be no effect on the pressure b. The pressure will increase c. the effect is impossible to determine d. The pressure will decrease

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**What will happen to
the pressure inside
a container of gas ...**

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recognized, adventure

as capably as

experience not quite

lesson, amusement, as

competently as pact

can be gotten by just

checking out a books

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Pressure is caused by
molecules hitting the
sides of a container or
other objects.

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A nonflexible container
is needed if the gas

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sample is going to have an internal pressure that is different from the external pressure. If a flexible container is used, the internal pressure and external pressure will always be the same because they are both pushing on the sides of the container equally.

Gas Quizlet: Packet Questions

Flashcards | Quizlet

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Temperature = 200 K
Experiment B (Heating
the gas) B1 B2 B3
Volume = 1 unit
Volume = 1 unit
Volume = 1 unit
External pressure = 1
atm External pressure
= 1 atm ...

Gas Variables How are the variables that describe a gas

...

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(Adding more gas) A1
A3A2 POGIL Chemistry
Activities - Flinn Gas
molecules keep their
distance from each
other and are in
constant motion.

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25 Gas Variables-S

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