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applied for a closed system, the heat transferred is given by:

$Q = \Delta U$  The change in internal energy is defined by Equation 8.3-6 of Felder & Rousseau:

$$\Delta U = \int_{T_1}^{T_2} C_p(T) dT$$

Substituting the heat capacity equation into the integral:

## **Chapter 8 Balances on Nonreactive Processes**

Date: Richard M. Felder,

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